

Acetoxyl Group Directed Cyclization Promoted by SmI₂: Directing Group Determined Stereocomplementarity

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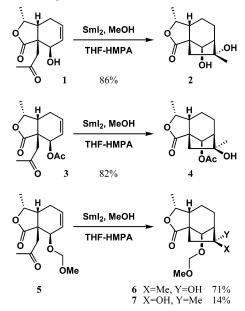
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Abstract: Complete reversal of diastereoselectivity was observed in the SmI₂-promoted ketyl-olefin coupling cyclizations of the hydroxy ketone or aldehyde and its acetate. For example, the stereodivergent synthesis of the epimeric five-membered-ring alcohols **2** and **4** has been accomplished through the SmI₂-induced ketyl-olefin coupling cyclizations of the δ -hydroxy ketone **1** and δ -acetoxy ketone **3**.

Samarium(II) iodide (SmI_2) has become an exceedingly reliable reagent for promoting the reductive coupling reactions difficult to accomplish by any other existing methodologies.¹ For example, the Barbier reactions, Reformatsky reactions, ketone-olefin reductive couplings, and pinacol couplings have been reported. Previous research from this laboratory established the powerful influence of a pendant hydroxyl group in directing the stereochemical course of the reductive cyclizations promoted by SmI₂.^{2,3} All of these hydoxyl-directed carboncarbon bond formation reactions proceed with good yields and provide entry into highly functionalized molecules with excellent stereocontrol, and the sense of the diastereoselectivity was in full accord with a chelationcontrol model. As a result of further investigations, we have now discovered another type of substrate-directable

SCHEME 1. Cyclization of Five-Membered Rings



annulation reaction promoted by SmI_2 . During the stereocontrolled cyclization, when the directing hydroxyl group was protected as its acetate ester, the SmI_2 -induced reductive coupling of the acetate was controlled by the acetoxyl group with a complete reversal of diastereoselectivity.^{4,5} In this paper, we describe this new and highly stereoselective cyclization mediated by SmI_2 .

The features of this new process are exemplified in the cyclization reactions of the γ -hydroxy ketone **1** and γ -acetoxy ketone **3** (Scheme 1). As previously reported, the hydroxyl-directed cyclization of **1** resulted in the exclusive generation of the diol **2**.^{2c} Remarkably, the SmI₂-induced cyclization of the corresponding γ -acetoxy ketone **3** proceeded with opposite diastereoselectivity, providing the alcohol **4** as the sole product.⁶ In these coupling products **2** and **4**, the stereochemistry about the new tertiary hydroxyl group is opposite. In contrast to these stereocomplementary cyclizations, the SmI₂-mediated coupling of the γ -methoxymethyoxy ketone **5** gave the epimeric mixture of the alcohols **6** and **7** (5:1).

As shown in Scheme 2, similar stereocomplementarity was also observed during the transformation of the carbohydrate templates, the δ -hydroxy aldehyde 8, and

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⁽¹⁾ Reviews: (a) Kagan, H. B.; Sasaki, M.; Collin, J. Pure Appl. Chem. 1988, 60, 1725. (b) Kagan, H. B. New J. Chem. 1990, 14, 453.
(c) Molander, G. A. Chem. Rev. 1992, 92, 29. (d) Molander, G. A.; Harris, R. H. Chem. Rev. 1996, 96, 307.

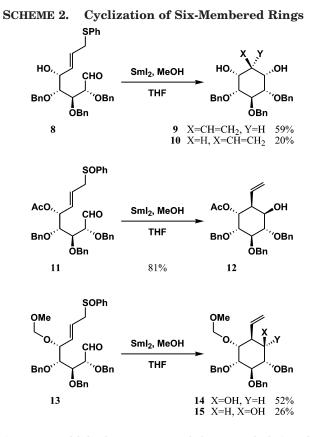
^{(2) (}a) Kan, T.; Matsuda, F.; Yanagiya, M.; Shirahama, H. Synlett
1991, 391. (b) Kito, M.; Sakai, T.; Yamada, K.; Matsuda, F.; Shirahama, H. Synlett
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1997, 479. (i) Matsuda, F.; Shirahama, H.; Matsuda, F.; Shirahama, H.; Matsuda, F.; Shirahama, H.; Matsuda, F. Angew. Chem. 2000, 112, 363; Angew. Chem., Int. Ed. 2000, 39, 355.

⁽³⁾ Recently, the SmI₂-mediated 5- and 6-*exo-trig* cyclizations stereocontrolled by hydroxyl groups have been reported by Molander and Losada. During these hydroxyl-directed transformations, the observed diastereoselectivity can be rationalized by the chelation-control model proposed by us: Molander, G. A.; Losada, C. P. J. Org. Chem. **1997**, 62, 2935.

⁽⁴⁾ In contrast, acetylation of the directing hydroxyl groups resulted in complete depression of the ability of the substrates to undergo the coupling reactions in several of the SmI_2 -mediated coupling cases already reported: ref 2b,d,i.

⁽⁵⁾ Complete stereocomplementarity was also observed in the intermolecular, SmI_2 -promoted ketone-olefin coupling chelationcontrolled by the α -hydroxyl and α -(alkoxycarbonyl)amino groups: (a) Kawatsura, M.; Dekura, F.; Shirahama, H.; Matsuda, F. Synlett **1996**, 373. (b) Matsuda, F.; Kawatsura, M.; Dekura, F.; Shirahama, H. J. Chem. Soc., Perkin Trans. 1 **1999**, 2371.

⁽⁶⁾ The relative stereochemistry of **4** was verified as follows. Removal of the acetyl group of **4** via basic hydrolysis gave the diol (the epimer of the diol **2** with regard to the tertiary hydroxyl group), of which the physical and spectral data were completely identical with those of the authentic sample having definite stereochemistry. For the synthesis of the authentic diol (the epimer of the diol **2**) see: Kan, T.; Oikawa, M.; Hosokawa, S.; Yanagiya, M.; Matsuda, F.; Shirahama, H. *Synlett* **1994**, 801, 805.



 δ -acetoxy aldehyde 11 prepared from methyl β -D-glucopyranoside into the six-membered carbocycles.⁷ As previously described, the cis-1,3-cyclohexanediols 9 and 10 (3:1) were exclusively obtained through the hydroxyldirected coupling mediated by SmI_2 of the δ -hydroxy aldehyde 8.2 On the other hand, the annulation reaction of the δ -acetoxy aldehyde 11 was controlled by the δ -acetoxyl group. When **11** was reacted with SmI₂, the self-terminating cyclization⁸ proceeded with complete stereochemical control to furnish the cyclohexanol 12 as the sole product.⁹ In the cyclization product 12, the relative stereochemistry between the hydroxyl and acetoxyl groups is trans on the newly formed six-membered ring. In contrast, the SmI₂-induced coupling of the δ -methoxymethyoxy aldehyde **13** gave a mixture of the epimeric alcohols 14 and 15 (2:1).

The observed *cis*-1,3-diol stereochemistry of the diols **2**, **9**, and **10** may be explained by assuming a chelationcontrol model as illustrated in Figure 1.^{2,3} Thus, chelation of the Sm(III) cation generated during the initial singleelectron transfer from SmI₂ to the γ -hydroxy ketone **1** or

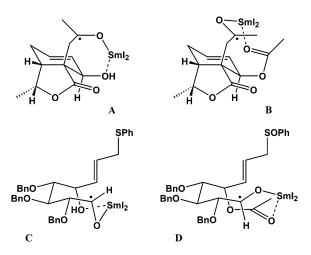


FIGURE 1. Chelation-control model.

 δ -hydroxy aldehyde **8** with the hydroxyl group produces the cyclic ketyl radicals **A** or **C**. The ketyl radial center adds to the olefinic part to form the carbon–carbon bond with complete stereoselectivity. On the other hand, the hypothesis that the ketyl-radical addition reaction proceeds involving the coordination of the carbonyl oxygen atom of the acetoxyl group to the Sm(III) cation attached to the ketyl radical, as shown in **B** or **D** during the cyclization of the γ -acetoxy ketone **3** or δ -acetoxy aldehyde **11**, may account for the observed dramatic reversal of diastereoselectivity.

As described in this paper, we are able to reveal the ability of the acetoxyl group to change the stereochemical course of the SmI₂-induced ketone-olefin cyclization as compared to the case where the hydroxyl group is free. From a synthetic point of view, the observed stereo-complementarity is highly advantageous because it enhances the possibilities of application to divergent synthesis of products with complementary stereochemistry about the newly formed hydroxyl group.⁵

Experimental Section¹⁰

Cyclization of 1. To a solution of 1 (70.0 mg, 0.263 mmol), MeOH (0.32 mL, 7.90 mmol), and HMPA (0.92 mL, 5.29 mmol) in THF (1.0 mL) cooled at -78 °C was added a 0.10 M THF solution of SmI₂ (13.2 mL, 1.32 mmol). The mixture was stirred at -78 °C for 2 h, quenched with saturated aqueous NH₄Cl, and extracted with AcOEt. The combined organic layers were washed with H₂O and brine, dried over Na₂SO₄, and concentrated in vacuo. The product was purified by silica gel column chromatography (AcOEt/hexane, 40:60) to give 2 (57.7 mg, 82%) as white crystals: mp 139–140 °C; $[\alpha]^{25}$ _D +38.4 (c 0.900, CHCl₃); IR (CHCl₃) 3400, 2950, 1750, 1730, 1450, 1350, 1250, 1020, 950 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 1.47 (3H, d, J = 7.0 Hz), 1.57 (3H, s), 1.73 (1H, d, J = 13.5 Hz), 2.08 (3H, s), 2.53 (1H, d)m), 2.64 (1H, d, J = 13.5 Hz), 4.70 (1H, dq, J = 8.2, 7.0 Hz), 5.35 (1H, s); MS (FAB-MS) m/z 269 [M⁺ + H]. Anal. Calcd for C14H20O5: C, 62.67; H, 7.51. Found: C, 62.82; H, 7.47.

Cyclization of 11. To a solution of **11** (85.0 mg, 0.136 mmol) in a mixture of THF (1.6 mL) and MeOH (1.0 mL) cooled at -78 °C was added a 0.10 M THF solution of SmI₂ (3.40 mL, 0.340 mmol). After being stirred at -78 °C for 3 h, the reaction mixture was quenched with saturated aqueous NH₄Cl and extracted with AcOEt. The combined organic layers were washed with H₂O and brine, dried over Na₂SO₄, and concentrated in vacuo. Purification

⁽⁷⁾ In the past decade, organic chemists have paid much attention to the conversion of carbohydrates to carbocycles. Now, a wide range of highly functionalized cyclopentane and cyclohexane derivatives have been made from sugar, the particular advantage being gained by the passing of chirality from the starting material to the product; review: Ferrier, R. J.; Middleton, S. *Chem. Rev.* **1993**, *93*, 2779.

⁽⁸⁾ Utilization of the allyl sulfoxide as a ketyl radical acceptor is essential for the SmI_2 -induced annulation reactions of the aldehydes **11** and **13** in contrast to that of the allyl sulfide **8**. Treatment of the corresponding allyl sulfides with SmI_2 affected the reduction of the aldehyde moieties to afford the primary alcohols. We reported the stereoselective cyclization mediated by SmI_2 using allyl sulfides and sulfones as a ketyl radical acceptor: Kan, T.; Nara, S.; Ito, S.; Matsuda, F.; Shirahama, H. J. Org. Chem. **1994**, 59, 5111.

⁽⁹⁾ The stereostructure of **12** was confirmed by 2D-COSY and 2D-NOESY experiments on **12**.

⁽¹⁰⁾ The reactions of compounds 5 and 13 were performed under the same conditions as for the transformations of 3 and 11.

JOC Note

by silica gel column chromatography (AcOEt/hexane, 40:60) gave **12** (55.3 mg, 81%) as a colorless oil: $[\alpha]^{25}{}_{\rm D}$ –32.5 (*c* 0.300, CHCl₃); IR (neat) 3410, 2920, 1745, 1655, 1460, 1375, 1255, 1090, 925 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 2.12 (3H, s), 2.93 (1H, br s), 3.55 (1H, t, *J* = 9.1 Hz), 3.76 (1H, dd, *J* = 3.5, 9.1 Hz), 3.85 (1H, t, *J* = 9.1 Hz), 3.98 (1H, dd, *J* = 5.4, 9.1 Hz), 4.52, 4.68, 4.70, 4.76, 4.93, 4.99 (each 1H, d, *J* = 11.0 Hz), 5.26 (1H, d, *J* = 18.1 Hz), 5.28 (1H, d, *J* = 10.0 Hz), 5.46 (1H, t, *J* = 3.5 Hz), 5.79 (1H, ddd, *J* = 7.6, 10.0, 18.1 Hz), 7.27–7.40 (15H, m); MS (EI) *m/z* 502 [M⁺]. Anal. Calcd for C₃₁H₃₄O₆: C, 74.08; H, 6.82. Found: C, 73.95; H, 6.98.

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